



$$1 \text{ Ca} \times 40.1 \text{ u} = 40.1 \text{ u}$$

$$4 \text{ C} \times 12.0 \text{ u} = 48.0 \text{ u}$$

$$6 \text{ H} \times 1.0 \text{ u} = 6.0 \text{ u}$$

$$4 \text{ O} \times 16.0 \text{ u} = \underline{64.0 \text{ u}}$$

$$158.1 \text{ u}$$



$$1 \text{ Sn} \times 118.7 \text{ u} = 118.7 \text{ u}$$

$$4 \text{ Cl} \times 35.5 \text{ u} = \underline{142.0 \text{ u}}$$

$$260.7 \text{ u}$$



$$5 \text{ N} \times 14.0 \text{ u} = 70.0 \text{ u}$$

$$4 \text{ C} \times 12.0 \text{ u} = \underline{48.0 \text{ u}}$$

$$118.0 \text{ u}$$



$$2 \text{ In} \times 114.8 \text{ u} = 229.6 \text{ u}$$

$$3 \text{ O} \times 16.0 \text{ u} = \underline{48.0 \text{ u}}$$

$$277.6 \text{ u}$$

SiS

$$\begin{array}{r} 1 \text{ Si} \times 28.1 \text{ u} = 28.1 \text{ u} \\ 1 \text{ S} \times 32.1 \text{ u} = \underline{32.1 \text{ u}} \\ 60.2 \text{ u} \end{array}$$

Sn(NO₂)₂

$$\begin{array}{r} 1 \text{ Sn} \times 118.7 \text{ u} = 118.7 \text{ u} \\ 2 \text{ N} \times 14.0 \text{ u} = 28.0 \text{ u} \\ 4 \text{ O} \times 16.0 \text{ u} = \underline{64.0 \text{ u}} \\ 210.7 \text{ u} \end{array}$$

Y₂O₃

$$\begin{array}{r} 2 \text{ Y} \times 88.9 \text{ u} = 177.8 \text{ u} \\ 3 \text{ O} \times 16.0 \text{ u} = \underline{48.0 \text{ u}} \\ 225.8 \text{ u} \end{array}$$

N₂O₃

$$\begin{array}{r} 2 \text{ N} \times 14.0 \text{ u} = 28.0 \text{ u} \\ 3 \text{ O} \times 16.0 \text{ u} = \underline{48.0 \text{ u}} \\ 76.0 \text{ u} \end{array}$$



$$\begin{array}{r} 2 \text{ Cu x } 63.5 \text{ u} = 127.0 \text{ u} \\ 1 \text{ Si x } 28.1 \text{ u} = 28.1 \text{ u} \\ 3 \text{ O x } 16.0 \text{ u} = \underline{48.0 \text{ u}} \\ 203.1 \text{ u} \end{array}$$



$$\begin{array}{r} 2 \text{ Co x } 58.9 \text{ u} = 117.8 \text{ u} \\ 3 \text{ Cr x } 52.0 \text{ u} = 156.0 \text{ u} \\ 12 \text{ O x } 16.0 \text{ u} = \underline{192.0 \text{ u}} \\ 465.8 \text{ u} \end{array}$$



$$\begin{array}{r} 2 \text{ Ag x } 107.9 \text{ u} = 215.8 \text{ u} \\ 1 \text{ Cr x } 52.0 \text{ u} = 52.0 \text{ u} \\ 4 \text{ O x } 16.0 \text{ u} = \underline{64.0 \text{ u}} \\ 331.8 \text{ u} \end{array}$$



$$\begin{array}{r} 3 \text{ Sn x } 118.7 \text{ u} = 356.1 \text{ u} \\ 2 \text{ P x } 31.0 \text{ u} = 62.0 \text{ u} \\ 8 \text{ O x } 16.0 \text{ u} = \underline{128.0 \text{ u}} \\ 546.1 \text{ u} \end{array}$$



$$\begin{array}{r} 2 \text{ Au x } 197.0 \text{ u} = 394.0 \text{ u} \\ 2 \text{ C x } 12.0 \text{ u} = 24.0 \text{ u} \\ 4 \text{ O x } 16.0 \text{ u} = \underline{64.0 \text{ u}} \\ 482.0 \text{ u} \end{array}$$



$$\begin{array}{r} 1 \text{ Pb x } 207.2 \text{ u} = 207.2 \text{ u} \\ 1 \text{ O x } 16.0 \text{ u} = \underline{16.0 \text{ u}} \\ 223.2 \text{ u} \end{array}$$



$$\begin{array}{r} 2 \text{ Na x } 23.0 \text{ u} = 46.0 \text{ u} \\ 1 \text{ S x } 32.1 \text{ u} = \underline{32.1 \text{ u}} \\ 78.1 \text{ u} \end{array}$$



$$\begin{array}{r} 1 \text{ N x } 14.0 \text{ u} = 14.0 \text{ u} \\ 5 \text{ H x } 1.0 \text{ u} = 5.0 \text{ u} \\ 1 \text{ O x } 16.0 \text{ u} = \underline{16.0 \text{ u}} \\ 35.0 \text{ u} \end{array}$$



$$1 \text{ Cu} \times 63.5 \text{ u} = 63.5 \text{ u}$$

$$1 \text{ S} \times 32.1 \text{ u} = 32.1 \text{ u}$$

$$4 \text{ O} \times 16.0 \text{ u} = \underline{64.0 \text{ u}}$$

$$159.6 \text{ u}$$



$$1 \text{ Al} \times 27.0 \text{ u} = 27.0 \text{ u}$$

$$3 \text{ N} \times 14.0 \text{ u} = 42.0 \text{ u}$$

$$6 \text{ O} \times 16.0 \text{ u} = \underline{96.0 \text{ u}}$$

$$165.0 \text{ u}$$