

Chemical Formula Intro

Write the chemical formula for the following compounds.

1. Copper II Chloride
2. Aluminum Oxide
3. Nickel III Bromide
4. Lithium Sulfide
5. Boron Oxide
6. Lead II Nitride
7. Beryllium Fluoride
8. Barium Chloride
9. Manganese IV Oxide
10. Iron III Sulfide
11. Silver Bromide
12. Niobium V Oxide
13. Platinum IV Nitride
14. Cadmium Phosphide
15. Chromium VI Selenide

Why do compounds such as numbers 2, 4 and 5 not contain Roman Numerals?

Chemically, what do compounds 1 thru 15 have in common?

When dealing with ionic charges, what does the lowest common multiple represent?