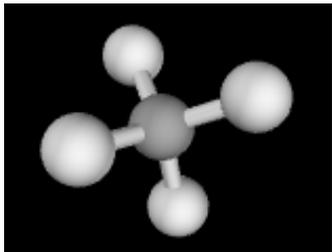
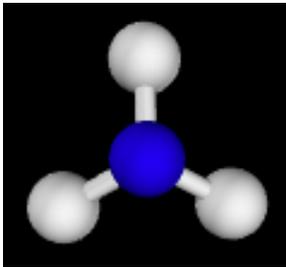
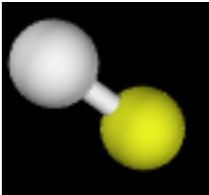
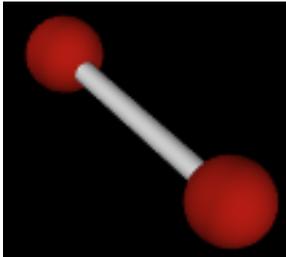


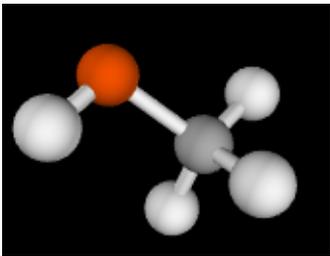
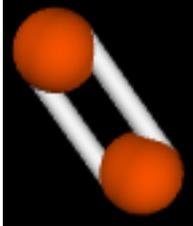
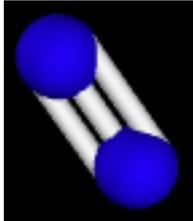
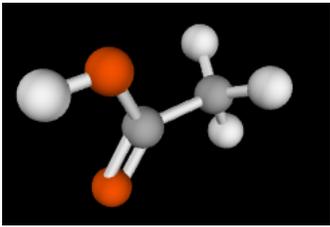
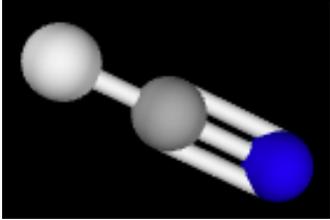
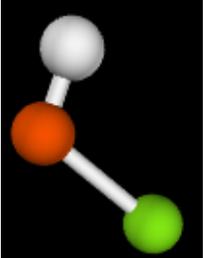
Build a Molecule Activity

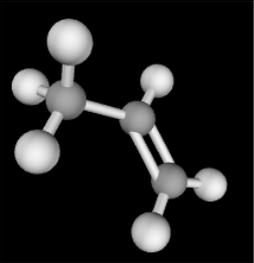
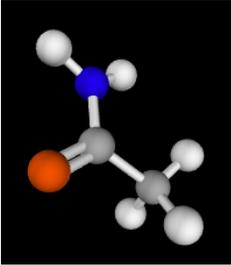
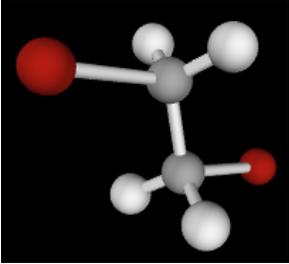
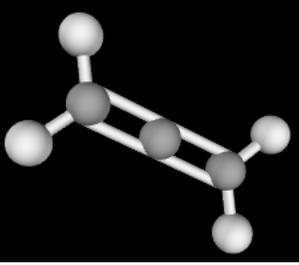
Name:

Class:

1. Using the LDMS homework 3 & 4, draw in the LDMS for each of the molecules.
2. Go to phet.colorado.edu/en/simulations/build-a-molecule click on the play symbol to open the activity.
3. Click on "Playground"
4. Use the "buckets" of atoms to build the the molecules in the chart. Click the arrows to access different atom combinations. Drag and drop the atoms into the space above the buckets to "build" molecules based on their LDMS.
5. Once the molecule is correctly built, its common name will appear. Write its common name.
6. Click "3D" to see its shape. Draw its shape.

Molecule	LDMS	Drawing of 3D molecule	Common name
CH_4 Domain: Molecular: Angle:	$\begin{array}{c} \text{H} \\ \\ \text{H}-\text{C}-\text{H} \\ \\ \text{H} \end{array}$		methane
NH_3 Domain: Molecular: Angle:	$\begin{array}{c} \text{H}-\overset{\cdot\cdot}{\underset{\cdot\cdot}{\text{N}}}-\text{H} \\ \\ \text{H} \end{array}$		ammonia
HF Domain: Molecular: Angle:	$\text{H}-\overset{\cdot\cdot}{\underset{\cdot\cdot}{\text{F}}}\cdot\cdot$		hydrogen fluoride
Br_2 Domain: Molecular: Angle:	$\cdot\cdot\overset{\cdot\cdot}{\underset{\cdot\cdot}{\text{Br}}}-\overset{\cdot\cdot}{\underset{\cdot\cdot}{\text{Br}}}\cdot\cdot$		molecular bromine

Molecule	LDMS	Drawing of 3D molecule	Common name
<p>CH₃OH</p> <p>Domain:</p> <p>Molecular:</p> <p>Angle:</p>	$\begin{array}{c} \text{H} \\ \\ \text{H}-\text{C}-\ddot{\text{O}}-\text{H} \\ \\ \text{H} \end{array}$		methanol
<p>O₂</p> <p>Domain:</p> <p>Molecular:</p> <p>Angle:</p>	$\ddot{\text{O}}=\ddot{\text{O}}$		molecular oxygen
<p>N₂</p> <p>Domain:</p> <p>Molecular:</p> <p>Angle:</p>	$\text{N}\equiv\text{N}$		molecular nitrogen
<p>CH₃COOH</p> <p>Domain:</p> <p>Molecular:</p> <p>Angle:</p>	$\begin{array}{c} \text{H} \quad \text{:O:} \\ \quad // \\ \text{H}-\text{C}-\text{C}-\ddot{\text{O}}-\text{H} \\ \\ \text{H} \end{array}$		acetic acid
<p>HCN</p> <p>Domain:</p> <p>Molecular:</p> <p>Angle:</p>	$\text{H}-\text{C}\equiv\text{N}:$		hydrogen cyanide
<p>HClO</p> <p>Domain:</p> <p>Molecular:</p> <p>Angle:</p>	$\text{H}-\ddot{\text{O}}-\ddot{\text{Cl}}:$		hypochlorous acid

Molecule	LDMS	Drawing of 3D molecule	Common name
<p>CH_3CHCH_2</p> <p>Domain:</p> <p>Molecular:</p> <p>Angle:</p>	$\begin{array}{c} \text{H} & \text{H} & \text{H} \\ & & \\ \text{H}-\text{C}-\text{C}=\text{C}-\text{H} \\ \\ \text{H} \end{array}$		<p>prop-1-ene</p>
<p>CH_3CONH_2</p> <p>Domain:</p> <p>Molecular:</p> <p>Angle:</p>	$\begin{array}{c} \text{H} & \text{O} \\ & \\ \text{H}-\text{C}-\text{C}-\text{N}-\text{H} \\ & \\ \text{H} & \text{H} \end{array}$		<p>acetamide</p>
<p>$\text{C}_2\text{H}_4\text{Br}_2$</p> <p>Domain:</p> <p>Molecular:</p> <p>Angle:</p>	$\begin{array}{c} \text{H} & \text{H} \\ & \\ \text{:Br}-\text{C}-\text{C}-\text{Br:} \\ & \\ \text{H} & \text{H} \end{array}$		<p>1,2-dibromoethane</p>
<p>CH_2CCH_2</p> <p>Domain:</p> <p>Molecular:</p> <p>Angle:</p>	$\begin{array}{c} \text{H} & & \text{H} \\ & & \\ \text{H}-\text{C}=\text{C}=\text{C}-\text{H} \end{array}$		<p>allene</p>