

Heisenberg Uncertainty Principle

Probability

Bohr	Schrodinger

- Energy Level Quantum Number (____)
 - _____ of energy level (cloud)
 - Value on periodic table from _____
 - Capacity = _____ rule
 - As n _____
 - _____ increases
 - _____ increases
 - _____ increases
- Sub Level Quantum Number (____)
 - The number of sub-levels in an _____ is _____
 - Names ____, ____, ____, and ____
- Orbital Quantum Number (____)
 - Space occupied by ____ electrons
- Spin Quantum Number (____)
 - In order for electrons to pair, they must have
 - _____ or _____