## Nuclear Decay Homework

Answer the following questions on a separate sheet of paper. Use complete sentences when necessary.

- 1) How does an unstable nucleus release energy?
- 2) What are the three main types of nuclear radiation?
- 3) What part of an atom undergoes change during radioactive decay?
- 4) How is the atomic number of a nucleus changed by alpha decay? By beta decay? By gamma decay?
- 5) How is the atomic mass number of a nucleus changed by alpha decay? By beta decay? By gamma decay?
- 6) Which of the three kinds of radiation described in this section is the most penetrating?
- 7) Complete and balance the equations for the following nuclear reactions.

$$^{140}_{58}\text{Ce} \Rightarrow ^{140}_{59}\text{Pr} + \underline{\qquad} (\underline{\qquad} \text{decay})$$

$$^{238}_{92}\text{U} \Rightarrow ^{234}_{90}\text{Th} + \underline{\qquad} (\underline{\qquad} \text{decay})$$

$$^{128}_{53}\text{I} \Rightarrow ^{124}_{51}\text{Sb} + \underline{\qquad} (\underline{\qquad} \text{decay})$$

$$^{211}_{84}\text{Po} \Rightarrow ^{211}_{85}\text{At} + \underline{\qquad} (\underline{\qquad} \text{decay})$$

- 8) Write the symbol and charge for each.
  - a. alpha particle
  - b. beta particle
  - c. gamma ray
- 9) How are the mass number and atomic number of a nucleus affected by the loss of the following?
  - a. Beta particle
  - b. Alpha particle
  - c. Gamma ray

Uranium-238 undergoes the following decay processes as it becomes Lead-206: alpha, beta, beta, alpha, alpha, alpha, alpha, beta, bet

Based on this cascade, fill in the blanks to track this transmutation process from U to Pb by taking the product of each decay and make it the starting point for the next decay.